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THE SULFINATE-SULFONE PUMMERER REARRANGEMENT

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Abstract: α -CH acidic sulfoxides with additional electron-withdrawing substitution at C- α afford generation of α sulfonyl thioethers during reaction with sulfinyl chlorides. Some reaction details as well as a mechanistic proposal are presented.

Introduction

The PUMMERER rearrangement is an acid catalyzed intramolecular redox reaction of CH acidic sulfoxides which leads to reduction at sulfur and to oxidation of α carbon¹. A suchlike formed OH-group can be easily replaced by other nucleophilic groups. Thus, conversion of DMSO (1) i.e. with acyl chlorides leads to acyloxy sulfonium chlorides which are in equilibrium with chloro sulfonium carboxylates presumably via tetracovalent sulfur intermediates². Since on the one hand conversion of DMSO with thionyl chloride yields directly chloromethyl methyl sulfide³ (2), and MT-sulfone 4 is easily available from DMSO, acetic acid anhydride and sodium 4-toluene sulfinate⁴ via Pummerer reaction, sulfinyl chlorides on the other hand are oxygenated to sulfonyl chlorides 6 by sulfoxides⁵:

$$\begin{array}{c} & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$$

Results

Additional enhancement of sulfoxide α-CH acidity by electron withdrawing groups rendered possible sulfone formation via sulfinate-sulfone Pummerer rearrangement⁶:

R1= acvi: R2= H or acvi

Because of high CH acidity of open-chain 1.3-dicarbonyl compounds application of the aforementioned scheme by use of two mols of sulfinyl chloride $\underline{8}$ yields the corresponding α sulfonyl thioethers $\underline{10}^{6,7}$. This reaction sequence, however, fails with particularly strong acidic cyclic 1.3-dicarbonyl compounds $\underline{11}$ because their bis-sulfur substituted derivatives $\underline{12}$ represent very efficient sulfenylating agents which attack all suitable nucleophiles present in the reaction mixture.

Additionally, intermediacy of <u>13</u> gives rise to formation of capto-dative substituted radicals <u>16</u> which dimerize to give dimers <u>17</u> and subsequent products (<u>18-20</u>):

The related geminal β -dithiofunctionalization of α -morpholino styrene⁹ is of particular importance for mechanistic interpretation of this type of Pummerer rearrangement because it suggests sulfonyl radical migration within an radical cation-radical-pair rather than sulfinate formation:

$$\begin{array}{c} Ph \\ C = CH_{2} \xrightarrow{+2.R^{4}SOCI} \\ R_{2}N & CI^{-} & S_{2}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ R_{2}N & CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ R_{2}N & CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\ CI^{-} & S_{3}^{+}O - S - R^{4} \end{array} \longrightarrow \begin{array}{c} Ph \\$$

In these cases a one-pot procedure affords directly i.e. MT-sulfone.

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